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Method for Operating a Sensor for Determining the Concentration
of Oxidizing Gases in Gas Mixtures

5 The invention relates to a method for operating a sensor for
determining the concentration of oxidizing gases, especially for
determining the nitrogen oxide concentration in exhaust gases of
internal combustion engines in accordance with the preamble of
claim 1.

State of the Art

10 Such a sensor is presented, for example, in EP 0 791 826 A1.

Electrical fields and electrical currents arise between the
individual electrodes and between the electrodes and the heater
which cause the measuring result to be incorrect. This occurs
because all electrodes of such a sensor are conductively
15 connected to the solid state electrolyte and the insulation layer
of the heater has a finite resistance and, accordingly, all
electrodes are connected to each other via electrically
conductive structures and are connected at high resistance to the
heater.

20 Advantages of the Invention

The method of the invention having the features set forth in
claim 1, in contrast, offers the advantage that the measuring
errors can be eliminated by active compensation or at least can
be minimized. The measurement errors arise because of the mutual
25 coupling of the electrodes via electrical fields and currents in
the solid state electrolyte as well as by the voltage drops
across the feed line resistances. It is possible to precisely
adjust the voltages on the electrodes without errors being made
incorrect by voltage drops on the electrode feed lines or because
30 of a mutual coupling of electrodes. This is made possible by

changing the voltages, which are applied to the electrodes in accordance with function, in dependence upon the currents which flow in the electrode feed lines and/or between the electrodes. It is especially advantageous that the adjustment is independent of the current intensity with which the individual electrodes are charged.

An advantageous embodiment provides that one adds voltages to the voltages applied to the electrodes. The voltages added correspond to a feedback of voltage components weighted with factors and these voltage components are proportional to the currents. Furthermore, the sliding mean values of the voltages and/or their derivation of higher order and/or their sliding mean values or linear combinations thereof can be fed back. These voltages are proportional to the currents and the mean values are formed by means of known electric circuit elements. In this way, it is also possible to eliminate capacitive couplings.

The adjustment of the voltage on the electrodes takes place in this case advantageously by changing these factors. These factors are increased until the system starts to oscillate because of the feedback. The oscillation arises when the fed back factor is ≥ 1 in magnitude and, at the same time, the phase is greater or equal to 180° . Then, the factors are reduced slightly but only so far that just no oscillation occurs anymore. In this way, almost all voltage drops, which arise at the electrode feed lines, as well as the voltage drops which arise because of a fictive resistance network within the solid state electrolytes, can be compensated.

Drawing

Further advantages and features of the invention are the subject matter of the following description as well as the

schematic representation of an embodiment of the invention.

The drawings show:

FIG. 1 is a schematic section view through a sensor, which is known from the state of the art, for determining oxides in gas mixtures;

FIG. 2 shows schematically a circuit arrangement, which is known from the state of the art, for a sensor shown in FIG. 1;

FIG. 3 is an embodiment of a circuit arrangement for a sensor shown in FIG. 1 which is suitable for carrying out the method of the invention; and,

FIG. 4 schematically shows the coupling of the voltages/currents in matrix form which lie across the electrodes of a sensor shown in FIG. 1.

Description of the Embodiments

An NOx double chamber sensor is shown in FIG. 1 and includes five electrodes, namely: an oxygen pump electrode 9 subjected to the exhaust gas; an oxygen pump electrode 7 mounted in a first chamber and essentially lying opposite to the oxygen pump electrode 9 subjected to the exhaust gas; an oxygen pump electrode 8 arranged in a second chamber 2; and NO pump electrode 10 mounted likewise in the second chamber 2; and, an air reference electrode 6 mounted in a third chamber 3.

The first chamber 1 is connected via a diffusion barrier 4 to the exhaust gas, the second chamber 2 is connected to the first chamber via a further diffusion barrier 5.

The third chamber 3 is connected to the atmosphere via a channel.

The oxygen pump electrodes 7 and 8 pump oxygen away from the first chamber 1 or from the second chamber 2. The external pump electrode 9 functions as a counter electrode.

Nitrogen oxides are pumped away by the NO pump electrode 10. All electrodes are arranged on an ion-conducting solid state electrolyte 20 which, for example, can be made of zirconium oxide and are electrically conductively connected therewith.

5 An insulated heater 11 is provided in order to heat up the sensor to the necessary operating temperature.

 An evaluation circuit functions to operate the sensor and this circuit makes various electrical voltages available and obtains the measurement signal from a current measurement. A
10 block circuit diagram of such a circuit, which is known from the state of the art, is shown schematically in FIG. 2. The three voltages for the oxygen pump electrodes (7, 8) as well as for the NO pump electrode 10 are generated by voltage
 references 31, 32, 33 and drivers 41, 42, 43 and are shifted by
15 the potential of the air reference. The oxygen pump electrodes 7, 8 lie in the first chamber 1 and in the second chamber 2. For the above, the voltage, which is outputted by the driver 40, is added to or subtracted from the voltage outputted
 by drivers 41, 42, 43 in adding elements 61, 62, 63 in a manner
20 known per se. The potential of the outer pump electrode 9 is adjusted via a two-point controller 50 until the voltage difference between the oxygen pump electrode 7 and the air reference electrode 6 corresponds to a pregivable desired value. The other electrode potentials are adjusted directly. The NO
25 pump current can be measured via a current-voltage converter 80 known per se and be outputted as a measurement signal.

 All electrodes are conductively connected to the solid state electrolyte 20 and the insulation layer of the heater 11 has a finite resistance. For this reason, all electrodes are connected
30 to each other via a conductance network and are connected at high

ohmage to the heater 11. The numerically largest conductances are shown in FIG. 1 schematically by the resistances R_p . Likewise, feed conductances of the conductive paths to the electrodes are present which are likewise shown schematically in FIG. 1 by resistances R_L .

The basic idea of the invention is to make possible the adjustment of the required voltages directly at the electrodes without the voltage drop across the feed line resistors R_L or the mutual coupling of the electrodes via the resistances R_p making these electrode voltages incorrect.

This is solved by a method for operating a sensor which is explained in combination with a circuit shown in FIG. 3. In the circuit shown in FIG. 3, those elements which are identical to those in the circuit shown in FIG. 2 have the same reference numerals so that, with reference to their description, reference is made to the presentation made to the circuit shown in FIG. 2. The circuit shown in FIG. 3 differs from the circuit shown in FIG. 2 in that circuit arrangements are provided by which the voltages U_{IPE} , U_{NO} , U_{O2} , which are applied to the electrodes 7, 8, 10, 9, are changeable in dependence upon the currents flowing in the measurement lines and/or between the electrodes. These circuit arrangements include current/voltage converters 100, 110, 120 and circuit elements (compensation branches) 201, 202, 203, 204, 205, 206 which are weighted with compensation factors K_1 , K_2 , K_3 , K_4 , K_5 , K_6 in such a manner that a component, which is proportional to the currents, is so fed back to the electrodes that the components, which are coupled in via the solid state electrolyte 20, and the feed losses are compensated. With a feedback of this kind, the potentials of the electrodes, which can be measured on the feed lines, are

dependent upon the currents in the solid state electrolyte 20 and in the feed lines. The currents in the solid state electrolytes 20 are not accessible to a measurement but result at every location from the linear combination of currents in the feed lines. The total system is viewed as being linearly electrical. Because of the linear combination of the currents at each location, one obtains voltages also at the locations of the electrodes which are linearly dependent upon the feed line currents. The feedback takes place in such a manner that first the factor K1 is increased stepwise until there is an oscillation because of the feedback. Then, the factor K1 is again reduced slightly until just no oscillation occurs. Correspondingly, and if still necessary, one can proceed with the additional factors K2 to K6. In this way, it is ensured that practically all disturbing influences because of the electrode feed lines and because of the resistances between the electrodes are eliminated. These resistances are present in the solid state electrolyte 20 and are disturbing. Additionally, the sliding mean values of the voltages formed by means of electrical circuit elements and/or their derivatives of higher order and/or their sliding mean values or linear combinations thereof can be fed back. The voltages are proportional to the current. In this way, not only ohmic but also capacitive couplings are eliminated.

FIG. 4 shows schematically the coupling matrix. The lines are formed by the currents of the electrodes I_{pump} electrode 7, I_{O2} pump electrode 8 and I_{NO} pump electrode 10. The current of the inner oxygen pump electrode I_{pump} electrode 7 is relatively large compared to the other two and has therefore significant influence on the electrode voltages U_{IPE} or on the pump electrode 7, U_{O2} on the pump electrode 8 and U_{NO} on the pump

